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download logixpro portable descarga gratis logixpro portable descarga gratis Various approaches have been taken to try to reduce the production of hydrogen cyanide (HCN) during the processing of coal. One approach involves oxidizing the coal prior to cracking, thereby reducing the amount of HCN which is formed. Another approach involves "gasification" of the coal. "Gasification" means the conversion of coal into carbon and hydrogen gases which may be burned to provide energy. However, there is a limit to the amount of energy that can be derived from coal. It has been estimated that the energy content of a given amount of coal may be increased by a factor of four when the coal is gasified to produce hydrogen and carbon gases as compared to coal which has been fired. Thus, further progress in gasification of coal can provide a significant improvement in energy output from the use of coal. Another process which is particularly applicable to the conversion of coal to produce a gas that can be used to generate energy is known as "hydrocracking." Hydrocracking involves the cracking of coal and coal derivatives by a catalytic reaction in the presence of hydrogen at elevated temperature and pressure. The preferred catalyst is a supported metal such as a supported nickel or cobalt or a supported platinum or palladium. Thus, the conversion of coal to produce gas can be effected by hydrocracking processes. Various impurities are contained in coal and coal derivatives. These impurities are of particular concern in the hydrocracking of coal to produce gas because of the deleterious effects that such impurities can have on the conversion process. Impurities which are particularly troublesome are those which can poison the catalytic activity of the hydrocracking process. Such impurities include nitrogen and sulfur containing compounds. It is known that coal itself does not contain substantial amounts of nitrogen or sulfur compounds. Rather, nitrogen and sulfur compounds are contaminants which are derived from the air in which the coal is stored or generated. In addition to nitrogen and sulfur containing compounds, other compounds are also present in coal which can poison the catalytic activity of the hydrocracking process. Such compounds include ash, and various organic compounds. For example, coal commonly contains ash which is composed of carbonates and sulfides. Carbonates such as sodium carbonate, sodium bicarbonate and calcium carbonate can poison the catalytic activity of the hydrocracking process. Sulfides can also poison the catalytic activity of the hydrocracking process, as can organic 2d92ce491b